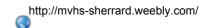
Friday, April 17/15 Science 122



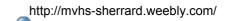
1. Questions?

Worksheet: Fluids - The Equation of Continuity Cutnell - Page 332 #56, 57, 58, 59 Worksheet - Fluids - Continuity and Bernoulli's Equations Worksheet - Fluids - Continuity and Bernoulli's Equations #2

- 2. Test Fluid Mechanics: *Moved from Friday to Monday.
- 3. Topic 5 Nuclear Physics
- 4. Review Atoms and Isotopes
- 5. Radioactive Decay (Alpha, Beta, Gamma) To Be Continued



Friday, April 17/15 Physics 122/121



- 1. Rewrites: Quiz U2-S1
- 2. Questions?

Worksheet - Circular Motion

Worksheet - Unbanked and Banked Curve Problems

- 3. Worksheet Kepler's Laws HW
- 4. Experiment 8.1 Kepler's Laws Page 49



Friday, April 17/15 Science 10

- 1. Quiz Monday
- 2. Lab Types of Reactions -> Complete
 - -> One Report Per Group
 - -> Due: Monday, April 18/15
- 3. Practice Quiz and Answer Key

Quiz - Compounds and Reactions

1. Be able to identify ionic and molecular compounds. ionic compounds - generally begins with a metal -> simple binary -> involving polyatomic ions

-> involving multivalent metals (roman numerals 1 to 10 required) 4 -> involving polyatomic ions and [u50+ multivalent metals molecular compounds - begin with a nonmetal or metalloid -> binary -> prefixes 1-10

-> diatomic molecules (hydrogen, nitrogen, oxygen, fluorine, chlorine, bromine and iodine) $-> S_8, P_4$ -> water, ammonia, hydrogen peroxide H20, NH3, H202

- 2. Be able to write the names and chemical formulas for ionic and Nall > Sodium chloride molecular compounds.
- 3. Be able to identify the reactants and products in a chemical

 $A + B \Rightarrow AB$ -> formation/synthesis/combination AB-> A +B -> decomposition A + β C > AC+ β -> single replacement/displacement
-> double replacement/displacement AB+ (D=AD+C) -> combustion hadro + 02 => H20+(2)

5. Be able to balance chemical reactions.

6 Be able to predict the products of chemical reactions

You'll need your 2 periodic tables.