## Formation/Synthesis Reactions

General Formula

yields
$$A + B \longrightarrow AB$$
substance substance compound
$$Reactants \qquad Product$$

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## **Decomposition Reactions**

General Formula

$$\begin{array}{cccc} & \text{yields} & & & \\ AB & \longrightarrow & A & + & B \\ & & \text{compound} & & \text{substance} & \text{substance} \\ & & & & & \\ Reactant & & & Products \end{array}$$

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Worksheet: Formation and Decomposition Reactions

$$\frac{\partial Ca}{\partial L} + O_2 \rightarrow \frac{\partial Ca}{\partial L} + O_2 \rightarrow \frac{\partial Ca}{\partial L} + O_3 \rightarrow \frac{\partial Ca}{\partial L} + O_4 \rightarrow \frac{\partial Ca}{\partial L} + O_5 \rightarrow \frac{\partial Ca}{\partial L} + O_6 \rightarrow \frac{\partial Ca}$$

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$$\frac{2|1_20_2}{Ca(0_3)} \frac{1}{20} + 0_2$$

$$\frac{1}{120} \frac{1}{20} \frac{1}{20} = 0$$

## Single Replacement Reactions

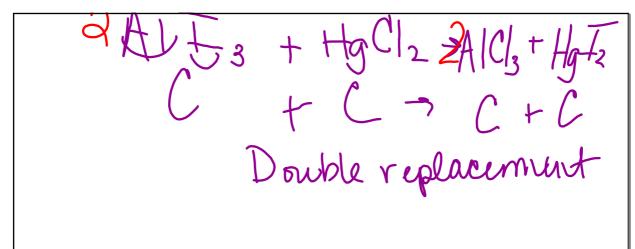
General Formula

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## Double Replacement Reactions

General Formula

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d AgNO3 + Ni > Ni(ND3)272Ag E C E Single replacement Science 10 - Grade 9 Chem Topics.docx

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