Physical and Chemical Changes (Page 28-30)

Matter can be grouped as metals and non-metals. When different metals are mixed together it is called an alloy

- 1. Give two examples of how matter changes
- 2. What is physical change and give 2 examples
- 3. Describe the different changes of state.
- 4. Are physical changes easy to reverse? Give an example.
- 5. What is a chemical change? Give 2 examples.
- 6. Are chemical changes easy to reverse? Give an example.
- 7. What are some clues as to whether a chemical change has occurred? If one of the clues is present, does it mean that a chemical change has definitely taken place?
- 8. Do question 2 on page 30.

Corrosion (Pages 34-35)

- 9. What is corrosion?
- 10. Is corrosion an example of a chemical or physical change? Explain
- 11. Describe rusting.
- 12. How can corrosion be prevented? Give 2 ways.
- 13. How is an oxide formed?

Combustion (Pages 38, 39)

- 14. What is combustion?
- 15. What are the 3 things needed for combustion?
- 16. What are fossil fuels? Give two examples.
- 17. How fid fossil fuels form?
- 18. Describe the word equation of fossil fuels. In this equation what are the reactants and what are the products.
- 19. What are hydrocarbons?
- 20. Why should you never operate a gas or charcoal barbecue inside a building?
- 21. Question 3 page 39.