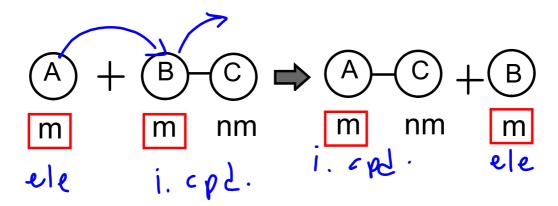
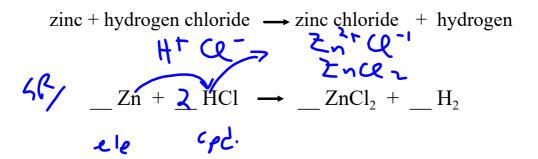
FA - Balancing Chemical Reactions (A)	
Name	
Balance each chemical equation below.	
<u> </u>	
1 <u>Pb</u> +O ₂ → <u>PbO</u>	
Reactants	Products
2. $\underline{CS_2 + }\underline{F_2} \rightarrow \underline{CF_4 + }\underline{S_8}$	
Reactants	Products
3. $\underline{CaBr_2 + Na_3P} \rightarrow \underline{Ca_3P_2 + NaBr}$	
Reactants	Products
$4. \ \underline{\hspace{1cm}} MgCl_2 + \underline{\hspace{1cm}} AgNO_3 \ \rightarrow \ \underline{\hspace{1cm}} Mg(NO_3)_2 \ + \underline{\hspace{1cm}} \underline{\hspace{1cm}}} \underline{\hspace{1cm}} \underline$	
	
Reactants	Products
5. <u>Li</u> +Cr ₂ (SO ₄) ₃ →Li ₂ SO ₄	+ Cr
Reactants	Products
6AlBO ₃ +K ₂ C ₂ O ₄ \rightarrow Al ₂ (C ₂ O ₄) ₅ +K ₃ BO ₃	
Reactants	Products

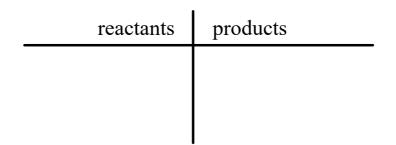
Single Replacement Reactions

General Format



Example #1





To Be Continued