

## Physics 112

Friday, February 14/20

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1. **Summative Assessment - Date:** Feb. 14/20
  2. **Independent Practice:U1-S1 - Vector Analysis**
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# Physics 122

Friday, February 14/20

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1. **IP - 2D Force Problems (Type II)**  
**IP - 2D Force Problems (Type II)**  
**IP - 2D Force Problems (Type III)**
  2. FA - 2D Force Problem (Type III)
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## Science 122

Friday, February 14/20

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1. FA - Double Lens Problem
  2. **Review: Mirrors and Lenses**
  3. SA - Optics -> Date: Wed., Feb. 19/20
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# Science 10

Friday, February 14/20

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1. **Summative Assessment: Periodic Table of Me, Myself and I**  
Due - Friday, Feb. 7/20  
5 Days Late Today

\*Get Sheets Initialed for Possible Re-assessment

2. **SA - Chemistry #1**  
- Date: Wed. Feb. 19/20
  3. FA - Standard Atomic Notation and Bohr-Rutherford Diagram  
- Checked in Class
  4. Worksheet - Bohr-Rutherford Diagrams: Atoms to Ions
  5. Periodic Table of Ions - To Be Continued
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6. Worksheet - Chemistry: Ions and Subatomic Particles
  7. Naming Monatomic Ions
  8. Nomenclature Worksheet #1 - Monatomic Ions
  9. FA - Atoms and Ions
  10. Handout - Ionic Compounds
  12. Simple Binary Ionic Compounds
  13. Nomenclature Worksheet #2 - Simple Binary Ionic Compounds

Science 10  
**Topics: SA - Chem #1**

1. chemistry
2. matter
3. types of properties: physical and chemical
4. types of changes: physical and chemical
5. atoms -> building blocks of matter
  - > three subatomic particles:  $p^+$ ,  $n$ ,  $e^-$
  - > locations of three subatomic particles
  - > electrically neutral:  $\#p^+ = \#e^-$
6. element
7. chemical symbols
8. periodic table of the elements - periods (rows)
  - groups/families (columns)
  - family and period names
  - location of metals, nonmetals and metalloids
  - characteristics of metals and nonmetals
9. atomic number = number of protons = # electrons (for atoms)
10. standard atomic notation -> mass # is atomic weight rounded to the nearest whole number
  - >  $\#N = \text{mass \#} - \text{atomic \#}$
11. Bohr-Rutherford Diagrams (for atoms)